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Day: Thursday Date: 10/16/2003 Time: 19:01:02

Inventor Name Search Result

Your Search was:

Last Name = HUA

First Name = XI

Application#	Patent#	Status	Date Filed	Title	Inventor Name
60364896	Not Issued Not	159	03/14/2002 10/16/2001	ENHANCED INHIBITION OF EPITHELIAL AND LEUKEMIA CELLS BY COMBINATION OF TGF-BETA AND RETINOIC ACID DIRECT TRANSMITTER	HUA, XIANXIN HUANG,
<u> </u>	Issued		10,10,2001	SELF-CALIBRATION TECHNIQUE	XINPING
60310475	Not Issued	159	08/08/2001	ENHANCED MANEUVERABILITY BY ACTIVE VORTEX CONTROL WITH MOVEABLE BLOWING JET	HUANG, XINGZHONG
60291148	Not Issued	159	05/15/2001	PARSING OF NESTED INTERNET ELECTRONIC MAIL DOCUMENTS	HUANG, XIAOLAN
60289990	Not Issued	159		METHODS FOR PREVENTING NEURONAL CELL DEGENERATION AND FOR STIMULATING AXON REGENERATION WITH LITHIUM	HUANG, XIZHONG
60283170	Not Issued	159	04/10/2001	LONG LASTING ANTI-BACTERIAL BIOCONJUGATES	HUANG, XICAI
60282967	Not Issued	159	04/11/2001	HYBRID NANOSTRUCTURED MATERIALS BASED ON II-VI SEMICONDUCTORS	11
60269973	Not Issued	159		MICROCHIP ELECTROSPRAY DEVICE AND COLUMN WITH AFFINITY ADSORBENTS AND USE OF THE SAME	HUANG, XIAN

60267832	Not Issued	159	02/09/2001	OPTIC NERVE REGENERATION IN BCL-2 TRANSGENIC MICE	HUANG, XIZHONG
60210890	Not Issued	159	06/09/2000	SURFACE MODIFICATION OF A POROUS POLYMER MONOLITH AND PRODUCTS THEREFROM	HUANG, XIAN
60200284	Not Issued	159	04/28/2000	USE OF RELAXIN TO TREAT DISEASES RELATED TO VASOCONSTRICTION	HUANG, XINFAN
60200001	Not Issued	159	04/27/2000	ACTIVE MATRIX ADDRESSED BISTABLE REFLECTIVE CHOLESTERIC DISPLAYS	HUANG, XIAO-YANG
60184808	Not Issued	159	02/24/2000	METHODS FOR DETERMINING SINGLE NUCLEOTIDE VARIATIONS	HUANG, XIAOHUA C.
60181408	Not Issued	159	02/09/2000	USE OF RELAXIN TO TREAT DISEASE RELATED TO VASOCONSTRICTION	
10610105	Not Issued	020	06/30/2003	CONTENT-BASED DYNAMIC PHOTO-TO-VIDEO METHODS AND APPARATUSES	HUA, XIAN-SHENG
10370314	Not Issued	020	02/19/2003	AUTOMATIC DETECTION AND SEGMENTATION OF MUSIC VIDEOS IN AN AUDIO/VIDEO STREAM	HUA, XIAN-SHENG
10368235	Not Issued	030		LEARNING-BASED AUTOMATIC COMMERCIAL CONTENT DETECTION	HUA, XIAN-SHENG
10286348	Not Issued	030	11/01/2002	SYSTEMS AND METHODS FOR AUTOMATICALLY EDITING A VIDEO	HUA, XIAN-SHENG
10222606	Not Issued	041	08/16/2002	ADDITIVE COMPOSITION FOR GEARBOX OIL	HUA, XIULING
09953798	Not Issued	030	09/17/2001	SYSTEM AND ELECTRONIC DEVICE FOR PROVIDING A MULTI-CARRIER SPREAD SPECTRUM SIGNAL	HUANG, XIAOJING
09949623	6504461	150	09/10/2001	OPEN MAGNET WITH RECESSED FIELD SHAPING COILS	HUANG, XIANRUI
09948756	Not Issued	161	09/10/2001	ALLELE DETECTION USING PRIMER EXTENSION WITH SEQUENCE-CODED IDENTITY TAGS	HUANG, XIAOHUA

<u>09941105</u>	Not Issued	030	08/28/2001	PARSING OF NESTED INTERNET ELECTRONIC MAIL DOCUMENTS	HUANG, XIAOLAN
09939809	6598391	150	08/28/2001	CONTROL FOR ELECTRO-HYDRAULIC VALVE ARRANGEMENT	HUANG, XIAODONG
09919455	Not Issued	161	07/31/2001	NOCATHIACIN ANTIBIOTICS PREPARED BY BIOTRANSFORMATION OR CHEMICAL METHODS	HUANG, XIAOHUA
<u>09884396</u>	6524242	150	06/19/2001	NON-CONTACT METHOD FOR MEASURING AMOUNT OF SEBUM OR OIL ON SUBSTRATE IN REAL TIME USING FLUORESCENCE DYE	HUA, XI YUAN
09884388	Not Issued	030	06/19/2001	MICROEMULSION FACIAL WASHES COMPRISING SPECIFIC OILS	HUA, XI YUAN
09837986	6627778	150	04/19/2001	SELECTIVE HYDROGENATION PROCESS FOR REMOVING C10 -C16 DIOLEFINS	HUANG,
<u>09836640</u>	Not Issued	161	the second of the second of the second	GRAPHIC CONTROLLER FOR ACTIVE MATRIX ADDRESSED BISTABLE REFLECTIVE CHOLESTERIC DISPLAYS	HUANG, XIAO-YANG
09836329	Not Issued	161	04/18/2001	OPERATING METHOD FOR ACTIVE MATRIX ADDRESSED BISTABLE REFLECTIVE CHOLESTERIC DISPLAYS	HUANG, XIAO-YANG
09836319	Not Issued	071	04/18/2001	ACTIVE MATRIX ADDRESSED BISTABLE REFLECTIVE CHOLESTERIC DISPLAYS	HUANG, XIAO-YANG
09818416	Not Issued	092]	HUANG, XIAOZHU
09810234 09764698	Not Issued	and the second s		METHOD FOR EXTRACTING OLEAGINOUS SUBSTANCES FROM GANODERMA LUCIDUM SPORES SEPARATION MEDIA,	HUANG, XIAO-NI HUANG, XIAN

				MULTIPLE ELECTROSPRAY NOZZLE SYSTEM AND METHOD	
09682589	Not Issued	071		BALANCED QUENCH PROTECTION CIRCUIT	HUANG, XIANRUI
09658070	6515433	150	09/11/2000	GAS DISCHARGE FLUORESCENT DEVICE	HUANG, XI
09657431	Not Issued	071	09/07/2000	LONG LASTING ANTIANGIOGENIC PEPTIDES	HUANG, XICAI
09657295	Not Issued	161	09/07/2000	METHODS AND COMPOSITIONS FOR PRODUCING LONG LASTING ANTINEOPLASTIC AGENTS	HUANG, XICAI
09623543	Not Issued	041		LONG LASTING ANTI-ANGIOGENIC PEPTIDES	HUANG, XICAI
09575081	Not Issued	093	05/19/2000	NOVEL ORGANIC ANION TRANSPORT PROTEINS	HUANG, XIN
09544025	Not Issued	071	04/05/2000	LIGANDS IN TREATING	HUANG, XIAOJIAN
09543789	6268152	150	04/06/2000		HUANG, XIAOHUA
09536841	Not Issued	041	03/27/2000	UNIVERSAL ARRAYS	HUANG, XIAOHUA
09510378	Not Issued	041	02/22/2000	ARRAYS OF NUCLEIC ACID PROBES ON BIOLOGICAL CHIPS	HUANG, XIAOHUA C.
09502799	Not Issued	160		METHOD FOR DIRECT METAL MAKING BY MICROWAVE ENERGY	HUANG, XIAODI
09502518	6277168	150	02/14/2000	METHOD FOR DIRECT METAL MAKING BY MICROWAVE ENERGY	HUANG, XIAODI
09497420	6355211	150		METHOD FOR MANUFACTURING HIGH PERFORMANCE COMPONENTS	HUANG, XIAODI
09468854	Not Issued	041		MICROBIAL TRANSFORMATION METHOD FOR THE PREPARATION OF AN EPOTHILONE	HUANG, XIAOHUA
09341399	6468744	150			HUANG,

				POLYMORPHISMS AND GENE COPY NUMBER	ХІАОНИА С.
<u>09194111</u>	6263227	150	01/19/1999	APPARATUS FOR IMAGING MICROVASCULAR BLOOD FLOW	HUANG, XIABING
08997021	Not Issued	161	12/23/1997	METHOD FOR IMPROVED PRECIOUS METAL RECOVERY	HUA, XI YUAN

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Application#	Patent#	Status	Date Filed	Title	Inventor Name
60408479	Not Issued	020	09/03/2002	TOTAL SYNTHESIS OF DIAZONAMIDE A	HUANG, XIANHAI
60403818	Not Issued	159	08/15/2002	AUTOMATED MLC LEAF POSITION LOCALIZATION AND LABELING WITH A NEW IMAGING PARAMETER DRIVEN TEMPLATE MATCHING ALGORITHM	HUANG, XIAOLEI
60403608	Not Issued	159	08/13/2002	LONG-LASTING GLYCOPEPTIDE DERIVATIVES	HUANG, XICAI
60355068	Not Issued	159	02/07/2002	SYNTHESIS OF SULFAMIDATES	HUANG, XIANHAI
60353265	Not Issued	159	02/01/2002	11 ~	HUANG, XIAOYAN
60348901	Not Issued	159	10/29/2001	LONG-LASTING VINORELBINE AND GEMCITABINE DERIVATIVES FOR TREATING CANCER	HUANG, XICAI
60344192	Not Issued	159	10/19/2001	AUTOMOBILE AS INTERMEDIATE MOBILE BASE STATION AND/OR REPEATER FOR WIRELESS COMMUNICATION BETWEEN MOBILE TERMINALS AND FIXED BASE STATION	HUANG, XI
60336950	Not Issued	159	10/26/2001	SAMPLE TUBE ARRAY	HUANG, XIAN
60329339	Not Issued	159		SELF-CALIBRATION TECHNIQUE	HUANG, XINPING
60327049	Not Issued	159	10/04/2001	HIGH-SPEED AND LOW POWER CONTENT ADDRESSABLE MEMORY (CAM) SENSING CIRCUITS	HUANG, XIAOHUA

60316305	Not Issued	159	08/31/2001	QUANTUM DOT AND QUANTUM DASH ACTIVE REGION DEVICES	HUANG, XIAODONG
60293174	Not Issued	159	05/25/2001	ACTIVE VORTEX CONTROL WITH MOVEABLE JET	HUANG, XINGZHONG
60274048	Not Issued	159	03/07/2001	LONG LASTING MATRIX METALLOPROTEASE INHIBITORS	HUANG, XICAI
60272617	Not Issued	159	03/01/2001	METHODS FOR STIMULATING AXON REGENERATION WITH LITHIUM	HUANG, XIZHONG
60272307	Not Issued	159	03/02/2001	TECHNIQUES FOR USING QUANTUM DOT ACTIVE REGIONS IN VERTICAL CAVITY SURFACE EMITTING LASERS	HUANG, XIADONG
60242216	Not Issued	159	10/20/2000	USE OF RELAXIN TO TREAT DISEASES RELATED TO VASOCONSTRICTION	HUANG, XINFAN
60231987	Not Issued	159	09/12/2000	METHODS AND COMPOSITIONS FOR PRODUCING LONG LASTING MATRIX METALLOPROTEASE INHIBITORS	HUANG, XICAI
60231182	Not Issued	159	09/07/2000	IMAGE SPATIAL SCALABILITY IN PIXEL DOMAIN AND DCT DOMAIN	HUANG, XIANHUA
60225114	Not Issued	159	08/14/2000	NOCATHIACIN ANTIBIOTICS PREPARED BY BIOTRANSFORMATION OR CHEMICAL METHODS	HUANG, XIAOHUA
60110192	Not Issued	159	11/30/1998	MULTI-METHOD SALES PIPELINE	HUANG, XIAOFEI JEFFREY
10205578	Not Issued	019	07/24/2002	ANALYSIS OF GENETIC POLYMORPHISMS AND GENE COPY NUMBER	HUANG, XIAOHUA C.
10202621	Not Issued	041	07/24/2002	CAM CELLS AND DIFFERENTIAL SENSE CIRCUITS FOR CONTENT ADDRESSABLE MEMORY (CAM)	HUANG, XIAOHUA
10202499	Not Issued	030	07/24/2002	HIGH SPEED AND LOW POWER SENSE CIRCUITS FOR CONTENT ADDRESSABLE MEMORY	HUANG, XIAOHUA
10140239	Not	030	05/07/2002	METHOD FOR	HUANG,

	Issued			MANUFACTURING CLAD COMPONENTS	XIAODI
10136565	Not Issued	030	05/01/2002	METHOD AND APPARATUS FOR COMPUTED TOMOGRAPHY IMAGING	HUANG, XIANGYU
10084628	Not Issued	071	02/25/2002	METHOD AND SYSTEM FOR SERVER-BASED OPERATIONS IN SERVER SYNCHRONIZATION WITH A COMPUTING DEVICE	HUANG, XIAO FEI
10084257	Not Issued	030	02/25/2002	METHOD AND SYSTEM FOR SERVER SYNCHRONIZATION WITH A COMPUTING DEVICE	HUANG, XIAO FEI
10079342	Not Issued	020	02/20/2002	PROTEIN STABILIZING AGENT	HUANG, XIAOLIN
10078843	Not Issued	041	02/19/2002	MICROCHIP ELECTROSPRAY DEVICE AND COLUMN WITH AFFINITY ADSORBENTS AND USE OF THE SAME	HUANG, XIAN
10036048	Not Issued	030	12/31/2001	METHOD AND APPARATUS FOR DETERMINING A CUSTOMER'S LIKELIHOOD OF REUSING A FINANCIAL ACCOUNT	HUANG, XIAO-MING
10035852	Not Issued	030	12/31/2001	METHOD AND APPARATUS FOR DETERMINING A CUSTOMER'S LIKELIHOOD OF PAYING OFF A FINANCIAL ACCOUNT	HUANG, XIAO-MING
10015261	Not Issued	030	12/12/2001	VEHICLE TELEMATICS RADIO OPERABLE FOR PROVIDING AND DISABLING DRIVING DIRECTIONS TO PRE-SELECTED DESTINATIONS	HUANG, XIAOPEI
10011161	Not Issued	030	12/05/2001	PROVIDING A PARTIALLY ENCRYPTED DATA PACKET IN A SPREAD SPECTRUM SIGNAL	HUANG, XIAOJING
10006885	Not Issued	093		ELECTRO-HYDRAULIC VALVE CONTROL SYSTEM AND METHOD	HUANG, XIAODONG
10005778	Not Issued	030		FILTER BANK AND RECEIVER FOR PROCESSING CONTINUOUS PHASE MODULATED SIGNALS	HUANG, XIAOJING
09973243	<u>6483563</u>	150	10/09/2001	BRIGHTNESS ENHANCEMENT	HUANG,

•				FOR BISTABLE CHOLESTERIC DISPLAYS	XIAO-YANG
09878495	Not Issued	161	06/11/2001	SURFACE MODIFICATION OF A POROUS POLYMER MONOLITH AND PRODUCTS THEREFROM	HUANG, XIAN
09861976	Not Issued	041	05/21/2001	LIGHTER WITH A LOCK-OFF SWITCH	HUANG, XINHUA
09861946	6443727	150	05/21/2001	LIGHTER WITH A LOCK-OFF MECHANISM	HUANG, XINHUA
09854937	6608409	150	05/15/2001	HIGH TEMPERATURE SUPER-CONDUCTING ROTOR HAVING A VACUUM VESSEL AND ELECTROMAGNETIC SHIELD AND AN ASSEMBLY METHOD	HUANG, XIANRUI
09810213	6440420	150	03/19/2001	METHOD FOR EXTRACTING OLEAGINOUS SUBSTANCES	HUANG, XIAO-NI
				FROM GERMINATION-ACTIVATED GANODERMA LUCIDUM SPORES	
09797405	Not Issued	030	03/01/2001		HUANG, XIAOLAN
09792413	Not Issued	161	02/23/2001	METHODS FOR DETERMINING SINGLE NUCLEOTIDE VARIATIONS	HUANG, XIAOHUA C.
09780752	Not Issued	041	02/09/2001	USE OF RELAXIN TREAT DISEASES RELATED TO VASOCONSTRICTION	HUANG, XINFAN
09780737	Not Issued	030	02/09/2001	DRIVE SCHEMES FOR GRAY SCALE BISTABLE CHOLESTERIC REFLECTIVE DISPLAYS UTILIZING VARIABLE FREQUENCY PULSES	HUANG, XIAO-YANG
09776768	Not Issued	061	02/06/2001	IDENTIFYING A BASE IN A NUCLEIC ACID	HUANG, XIAOHUA
<u>09731414</u>	6342578	150	12/06/2000	1	HUANG, XIAOYAN
09682880	Not Issued	030		DESIGN METHOD	HUANG, XIANRUI
09519090	6582908	150	03/06/2000	OLIGONUCLEOTIDES	HUANG,

					XIAOHUA C.
09510014	6274212	150	02/22/2000	METHOD TO DECREASE THE ACETALDEHYDE CONTENT OF MELT-PROCESSED POLYESTERS	HUANG, XIAOYAN
<u>09365695</u>	6316601	150	08/02/1999	ANTIBODIES SPECIFIC FOR B6 INTEGRINS	HUANG, XIAOZHU

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Inventor Name Search Result

Your Search was:

Last Name = VAN GORKOM First Name = LEONARD

Application#	Patent#	Status	Date Filed	Title	Inventor Name
09884396	6524242	150	06/19/2001	NON-CONTACT METHOD	VAN GORKOM,
					LEONARD
				OF SEBUM OR OIL ON	
				SUBSTRATE IN REAL TIME	
				USING FLUORESCENCE DYE	
<u>09884395</u>	<u>6475144</u>	150	06/19/2001	NON-CONTACT METHOD	VAN GORKOM,
					LEONARD
				OF SKIN SEBUM OR OIL IN	
				REAL TIME USING FIBER	
				OPTIC PROBE	
09884388	Not Issued	030	06/19/2001	MICROEMULSION FACIAL	VAN GORKOM,
				ll	LEONARD
			t till er er har er ha. he. he.	SPECIFIC OILS	
<u>08668151</u>	<u>5731277</u>	150	06/21/1996	AUTOMATIC DISHWASHING	VAN GORKOM,
				COMPOSITIONS	LEONARD
				CONTAINING ALUMINUM	
				TETRAHYDROXIDE	

Inventor Search Completed: No Records to Display.

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Your Search was:

Last Name = ARONSON First Name = MICHAEL Day: Thursday
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	Application#	Patent#	Status	Date Filed	Title	Inventor Name	
	60237073	Not Issued	159	09/30/2000		ARONSON, MICHAEL	
4	<u>29040064</u>	D376148	150		ELECTRONIC VOICE ANNUNCIATOR	ARONSON , MICHAEL L.	
X	10425206	Not Issued	030	04/29/2003	WET-SKIN TREATMENT COMPOSITIONS	ARONSON, MICHAEL PAUL	
	10340468	Not Issued	030	01/10/2003	METHODS OF CLEANSING, MOISTURIZING AND REFRESHING USING MULTIPHASE BARS HAVING ARTISAN-CRAFTED APPEARANCE	ARONSON, MICHAEL PAUL	
	10340457	Not Issued	030	01/10/2003		ARONSON, MICHAEL PAUL	
	10340153	Not Issued	030	01/10/2003	EXTRUDED MULTIPHASE BARS	ARONSON, MICHAEL PAUL	
		Not Issued	095	01/16/2002	COMPOSITIONS	ARONSON, MICHAEL PAUL	
	10050238	Not Issued	041	01/16/2002		ARONSON, MICHAEL PAUL	
	10034295	Not Issued	030		METHOD AND APPARATUS FOR DETECTING OPTIMUM LENS FOCUS POSITION	ARONSON, MICHAEL D.	
1	09884388	Not Issued	030	06/19/2001	WASHES COMPRISING SPECIFIC	ARONSON, MICHAEL PAUL	

09859862	Not Issued	041	05/17/2001	WET-SKIN TREATMENT COMPOSITIONS	ARONSON, MICHAEL PAUL
09859849	Not Issued	061	05/17/2001	METHOD OF ENHANCED MOISTURE OR REDUCED DRYING USING WET-SKIN TREATMENT COMPOSITIONS	ARONSON, MICHAEL PAUL
09796139	6521573	150	02/28/2001	PERSONAL CLEANSING COMPOSITIONS PROVIDING RINSABILITY AND DRAGGY WET SKIN FEEL	ARONSON, MICHAEL PAUL
09643142	6429177	150	08/22/2000	SEPARATING MULTI-PHASE PERSONAL WASH COMPOSITION IN A TRANSPARENT OR TRANSLUCENT PACKAGE	ARONSON, MICHAEL PAUL
09617777	Not Issued	092	07/17/2000	LAN PHONE SYSTEM WITH AUTOMATIC FALLBACK FOR POWER OR NETWORK FAILURE	ARONSON, MICHAEL
09559214	6218348	150	04/26/2000	PROCESS OF MAKING SOAP BAR WITH ENHANCED SKIN BENEFITS COMPRISING ADDING SALTS OF SPECIFIC PROTIC ACID	ARONSON, MICHAEL PAUL
09558821	Not Issued	041	04/26/2000	METHOD OF CLEANSING SKIN AND IMPROVING SKIN CARE IN BAR COMPRISING SOAP, FATTY ACID AND POLYALKYLENE GLYCOL	MICHAEL
09558810	6342470	150	04/26/2000	BAR COMPRISING SOAP, FATTY ACID, POLYALKYLENE GLYCOL AND PROTIC ACID SALTS IN CRITICAL RATIOS AND PROVIDING ENHANCED SKIN CARE BENEFITS	ARONSON, MICHAEL PAUL
09523131	6399045	150	03/10/2000	LIQUID SUNSCREEN COMPOSITIONS WHICH BOTH DEPOSIT AND LATHER WELL	ARONSON, MICHAEL PAUL
09252711	6265368	150	02/18/1999	II .	ARONSON , MICHAEL PAUL
08970736	6128673	150		METHOD AND APPARATUS FOR COMMUNICATION AND TRANSLATION OF A PLURALITY OF DIGITAL PROTOCOLS	ARONSON , MICHAEL D.

08903967	5760407	150	07/31/1997	DEVICE FOR THE IDENTIFICATION OF ACNE, MICROCOMEDONES, AND BACTERIA ON HUMAN SKIN	ARONSON , MICHAEL P.
08900654	Not Issued	161	07/25/1997	APPARATUS FOR USE IN A TELEPHONE CALL PROCESSING SYSTEM	ARONSON , MICHAEL D.
08898237	5961992	150	07/22/1997	BENEFIT AGENT COMPOSITIONS COMPRISING MIXTURES OF ALPHA-HYDROXY ESTERS	ARONSON , MICHAEL PAUL
08703747	5759969	150		PROCESS FOR MAKING AQUEOUS SOLUTION COMPOSITIONS COMPRISING POLYMER HYDROGEL COMPOSITIONS	ARONSON , MICHAEL P.
08412803	Not Issued	166	03/29/1995	LIQUID CLEANSER COMPOSITIONS COMPRISING POLYMER HYDROGEL COMPOSITIONS	ARONSON , MICHAEL P.
08411117	Not Issued	161	03/27/1995	CAPSULES WITH STRUCTURING AGENTS	ARONSON , MICHAEL P.
08366362	Not Issued	161	12/29/1994	TRANSFERRING DATA FROM CACHE MEMORY TO MAIN MEMORY	ARONSON , MICHAEL D
08152114	5498378	150	11/12/1993	PROCESS FOR PREPARING CAPSULES WITH STRUCTURING AGENTS	ARONSON , MICHAEL PAUL
08150701	5434069	150	11/12/1993	CAPSULE COMPRISING OIL SURROUNDING HYDROPHOBIC OR HYDROPHILIC ACTIVE AND POLYMERIC SHELL SURROUNDING OIL	ARONSON , MICHAEL P.
07875914	Not Issued	161	04/29/1992	11 ·	ARONSON , MICHAEL P.
	***			HYDROPHILIC WATER SOLUBLE POLYMER PHYSICALLY OR CHEMICALLY	
				ATTACHED TO A HYDROPHOBIC POLYMER CORE	
07875872	Not Issued	161	04/29/1992	HEAVY DUTY LIQUID DETERGENT COMPOSITIONS COMPRISING ENCAPSULATING POLYMER SYSTEM	ARONSON , MICHAEL P.
07580695	5066749	150		HYDROPHOBICALLY-MODIFIED POLYCARBOXYLATES AND	ARONSON , MICHAEL P.

C. C				PROCESS FOR THEIR PREPARATION	
07575270	5089163	150	08/30/1990	ENZYMATIC LIQUID DETERGENT COMPOSITION	ARONSON , MICHAEL P.
07566717	D337484	150	08/13/1990	SERVING TRAY	ARONSON , MICHAEL D.
07563451	5073285	150	08/06/1990	STABLY SUSPENDED ORGANIC PEROXY BLEACH IN A STRUCTURED AQUEOUS LIQUID	ARONSON , MICHAEL
07534780	5073292	150	06/07/1990	HEAVY DUTY LIQUID	ARONSON,
200000000000000000000000000000000000000				DETERGENT COMPOSITIONS CONTAINING ENZYMES	MICHAEL P.
XXXXXX				STABILIZED BY QUATERNARY	
100000000000000000000000000000000000000				NITROGEN SUBSTITUTED PROTEINS	
07455875	5045225	250	12/18/1989	SELF HYDROPHOBING SILICONE/HYDROCARBON ANTIFOAM COMPOSITIONS	ARONSON , MICHAEL P.
07364946	4992194	250	06/12/1989	STABLY SUSPENDED ORGANIC	ARONSON,
				PEROXY BLEACH IN A	MICHAEL
				STRUCTURED AQUEOUS LIQUID	
07305878	4908150	150	02/02/1989	STABILIZED LIPOLYTIC ENZYME-CONTAINING LIQUID DETERGENT COMPOSITION	ARONSON , MICHAEL P.
07304394	4959179	150	01/30/1989	STABILIZED ENZYMES LIQUID	ARONSON,
300				DETERGENT COMPOSITION CONTAINING LIPASE	MICHAEL P.
				PROTEASE	
07293725	4938888	150	01/05/1989	DETERGENT SHEET WITH ALKYL POLYGLYCOSIDE	ARONSON , MICHAEL P.
07292692	4917811	150	01/03/1989	COMPOSITION BLEACH COMPOSITIONS AND	ARONSON,
01232032	721/011	130	01/03/1707	Hall Notice (1919) - Illinoi Marketia, yr ei yr rafarau afi ffactiau a raceagau'r fyddiol, a addul	MICHAEL P.
07292300	Not Issued	166	12/30/1988	SELF HYDROPHOBING SILICONE/HYDROCARBON ANTIFOAM COMPOSITIONS AND DETERGENT COMPOSITIONS CONTAINING SAME	ARONSON , MICHAEL P.
07285466	Not Issued	161		STABILIZED GRANULAR BLEACH COMPOSITIONS	ARONSON , MICHAEL P.
07259072	4992212	150		ALKALINE LIGHT DUTY LIQUID DETERGENTS THAT ARE NON-STAINING TO ALUMINUM	ARONSON , MICHAEL P.

07246836	Not Issued	168	1 (1 (1 (1 (1 (1 (1 (1 (1 (1 (1 (1 (1 (1		ARONSON , MICHAEL P.
07183512	4877544	250		11	ARONSON , MICHAEL P.
07040386	Not Issued	166		1 (1997), 1997 − 199	ARONSON , MICHAEL P.
06931361	4762637	150		ENCAPSULATED BLEACH PARTICLES FOR MACHINE DISHWASHING COMPOSITIONS	ARONSON , MICHAEL
06103768	4606913	150	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1		ARONSON , MICHAEL P.

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Last Name = ZHU

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Application#	Patent#	Status	Date Filed	Title	Inventor Name
09884396	6524242	150	06/19/2001	NON-CONTACT METHOD	ZHU, ZHENHE
				FOR MEASURING AMOUNT	
	in guite sur ex			OF SEBUM OR OIL ON	
				SUBSTRATE IN REAL TIME	
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<u>09884395</u>	6475144	150	06/19/2001	NON-CONTACT METHOD	ZHU, ZHENHE
100				FOR MEASURING AMOUNT	
1.1				OF SKIN SEBUM OR OIL IN	
50000000000000000000000000000000000000				REAL TIME USING FIBER	
				OPTIC PROBE	
09884388	Not Issued	030	06/19/2001	MICROEMULSION FACIAL	ZHU, ZHENHE
				WASHES COMPRISING	
				SPECIFIC OILS	

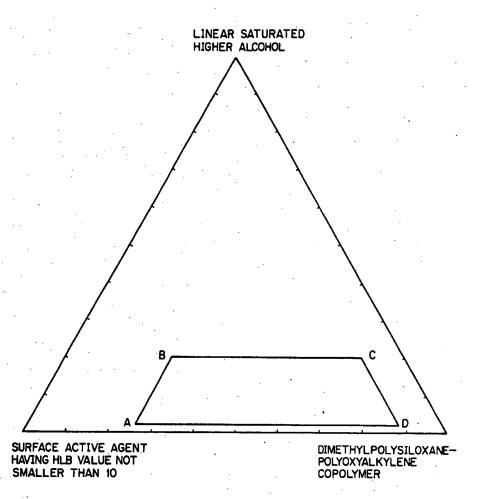
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United States Patent [19] 4,675,179 Patent Number: [11] Suzuki et al. Jun. 23, 1987 Date of Patent: [54] COSMETIC EMULSION AND METHOD FOR 4,554,369 11/1985 Hill et al. 424/59 MAKING THE SAME FOREIGN PATENT DOCUMENTS [75] Inventors: Toshiyuki Suzuki, Ichikawa; Akira 0022712 3/1981 Japan 424/70 Tsukada, Kodaira; Masanobu Kai, 0086113 7/1981 Japan 424/70 Funabashi, all of Japan 803289 10/1958 United Kingdom 514/63 [73] Assignee: Kao Corporation, Tokyo, Japan Primary Examiner—Dale R. Ore Attorney, Agent, or Firm-Oblon, Fisher, Spivak, [21] Appl. No.: 701,518 McClelland & Maier [22] Filed: Feb. 14, 1985 **ABSTRACT** Foreign Application Priority Data [30] A cosmetic emulsion of low viscosity which comprises Mar. 16, 1984 [JP] Japan 59-50547 (a) 0.2 to 5 wt % of an emusifier, (b) 0.5 to 10 wt % of an oil and (3) 85 to 99.3 wt % of a water phase. The Int. Cl.4 A61K 7/36; A61K 7/38 emusifier (a) consists essentially of a dimethylpolysilox-[52] U.S. Cl. 424/67; 424/68; 514/941 ane-polyalkylene copolymer, a surface active agent having an HLB value not smaller than 10, and a linear, [58] Field of Search 514/63, 941; 424/70, 424/66, 68, 67, 63 saturated higher alcohol having from 12 to 22 carbon atoms. The relative ratios of the dimethylpolysiloxane-[56] References Cited polyoxyalkylene copolymer, the surface active agent **U.S. PATENT DOCUMENTS** and the higher alcohol lie within the polygon bounded by the points of a ternary composition diagram of the 3,641,239 2/1972 Mohrlok 514/63 3,655,865 4/1972 Murphy 514/63 annexed figure. 4,254,104 3/1981 Suzuki 514/63 When the emulsion is applied to the skin, it gives re-4,264,586 4/1981 Callingham et al. 424/68 freshing feeling to the skin without any stickiness. It 4,268,499 5/1981 Keil 424/68 also gives appropriate moisture retentivity to the skin. 4,311,695 1/1982 Starch 514/63 4,423,032 12/1983 Abe et al. 514/63 4,423,041 12/1983 Chem et al. 514/63 6 Claims, 1 Drawing Figure

FIGURE



COSMETIC EMULSION AND METHOD FOR MAKING THE SAME

BACKGROUND OF THE INVENTION

(i) Field of The Invention

This invention relates to stable, low viscosity cosmetic emulsions which have the skin refreshed upon application thereof without becoming sticky after the application and which have appropriate moisture retentivity. The invention also relates to a method for making such emulsions.

(ii) Description of the Prior Art

In general, cosmetic emulsions consist of water, oils and emulsifiers and, when applied to the skin, they form 15 an oil film on the surface of the skin. By the coverage with the oil film, a percutaneous water loss is so suppressed that moisture in the horny layer is appropriately retained. The moisture-retentive effect depends on the type and amount of the oil, and thus the type and 20 amount of the oil are suitably controlled according to the purposes.

In order to attain less stickiness and refreshness to the touch by the use of emulsions, it is necessary to reduce the amount of oils and use oils which are less sticky.

The polysiloxane of the general formula (II), (III) or (IV) indicated hereinafter (hereinafter referred to silicone oil) is able to form a uniform thin film on the skin and is suitable as an oil imparting the stickness-free smoothness to the touch. However, the silicone oil is 30 rather poor in compatibility with ordinary surface active agents and other oils and has very poor ability of being emulsified, which makes it very difficult to obtain uniform, fine emulsions.

Better stability against coalescence and creaming is 35 formula (II), (III) or (IV) otained when emulsified particles are more uniform and finer. This tendency becomes more pronounced particularly when the viscosity of the system is low. The state of emulsification takes part in the appearance or whiteness of an emulsion. In this sense, optimal results are obtained when the particle size is in the range of from 0.2 to 2 μ m. At a smaller particle size, there is obtained a transparent, pale microemulsion. Over the above range, a semitransparent, greyish emulsion is obtained.

In order to further impart refreshness at the time of 45 application, the cosmetic system is preferred to contain ethyl alcohol. In general, addition of ethyl alcohol results in a considerable lowering of emulsion stability of the system.

The emulsion system having such properties as described above should satisfy the following requirements: ethyl alcohol is contained in an aqueous phase; an oil being emulsified is mainly composed of silicone oil and is contained in small amounts; and the system is low in viscosity and has emulsified particles which are 55 fine and uniform in size. However, known emulsification techniques cannot satisfy all the above requirements at the same time.

SUMMARY OF THE INVENTION

Under these circumstances, the present inventors have made intensive studies to obtain cosmetic emulsions for which the objects of the invention can be achieved. As a result, it has been found that when a mixture of a specific type of dimethylpolysiloxane-65 polyoxyalkylene copolymer, a surface active agent having an HLB value not smaller than 10, and a linear, saturated higher alcohol having from 12 to 22 carbon

atoms is used as an emulsifier, to which an aqueous phase comprising ethyl alcohol in a predetermined concentration is added under agitation in a predetermined range of temperature thereby forming one phase region of a gel or a highly viscous liquid, there can be obtained an emulsion with a low viscosity which is fine, uniform and stable and which contains the silicone oil as the main component. In addition, the emulsion composition satisfies the requirements for the cosmetic emulsion. The present invention is based on the above finding.

According to one feature of the invention, there is provided a cosmetic emulsion which comprises:

(a) 0.2 to 5 wt% of an emulsifier consisting essentially of (1) a dimethylpolysiloxane-polyoxyalkylene copolymer represented by the general formula (I)

$$\begin{array}{c|c} CH_{3} & CH_{3} & CH_{3} \\ CH_{3} - SiO & SiO & SiO & SiO & SiO & CH_{3} \\ CH_{3} & CH_{3} & CH_{3} & CH_{3} \\ CH_{3} & CH_{3} & CH_{3} & CH_{3} \\ \end{array}$$

in which a represents an integer of from 10 to 25, b is an integer of from 25 to 35, m is an integer of from 60 to 80, and n is an integer of from 3 to 8, (2) a surface active agent having an HLB value not smaller than 10, and (3) a linear, saturated higher alcohol having from 12 to 22 carbon atoms:

(b) 0.5 to 10 wt% of an oil comprising 90% or more of at least one polysiloxane represented by the general formula (II), (III) or (IV)

$$\begin{array}{c|c} CH_3 & CH_3 & CH_3 \\ CH_3 - SiO & SiO & SiO \\ CH_3 & CH_3 & CH_3 \\ CH_3 & CH_3 & CH_3 \\ \end{array}$$

in which x represents an integer of from 4 to 100, z is an integer not smaller than 1, y+z is a value from 1 to 100, and 1 is an integer of from 2 to 5; and

(c) 85 to 99.3 wt% of a water phase comprising 60 to 100 wt%, based on the total water phase, of an ethyl alcohol aqueous solution in which an ethyl alcohol to water ratio by weight is 50:50 to 2:98.

Another feature of the invention resides in provision of a method of making the cosmetic emulsion.

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BRIEF DESCRIPTION OF THE DRAWING

The sole FIGURE is a ternary composition diagram showing relative ratios of dimethylpolysiloxane-polyoxyalkylene copolymer, a surface active agent having an HLB value not smaller than 10, and a linear, saturated higher alcohol having 12 to 22 carbon atoms, which constitute an emulsifier of a cosmetic emulsion of the invention.

DETAILED DESCRIPTION OF THE INVENTION AND PREFERRED EMBODIMENTS

The dimethylpolysiloxane-polyoxyalkylene copolymer of (1), which is one component of the emulsifier (a) 15 of the cosmetic emulsion according to the invention, should preferably have a clouding point, as an aqueous solution thereof, of 25° to 40° C. If the clouding point of the aqueous solution exceeds 50° C., one phase emulsion cannot be formed during the emulsification and the 20 emulsified particles of the emulsion become coarse. On the other hand, when the clouding point is lower than 20° C., it does by no means serve as a surface active agent and thus no emulsification proceeds. The surface active agent (2) may be those agents having an HLB 25 value not smaller than 10, which may be used singly or in combination. In particular, surface active agents having an HLB value of from 11 to 16 are preferred. Specific examples of the surface active agents include polyoxyethylene alkyl ethers, polyoxyethylene and fatty 30 acid esters, fatty acid esters of polyoxyethylene and sorbitan, fatty acid esters of polyoxyethylene and glycerine, polyoxyethylene hardened castor oils, polyglycerine and fatty acid esters, sucrose and fatty acid esters, and the like.

The linear, saturated higher alcohols (3) should have from 12 to 22 carbon atoms, preferably 14 to 18 carbon atoms. When the number of carbon atoms are below 10, the action of stabilizing a material being emulsified lowers to a substantial extent with offensive odor. When the 40 number of carbon atoms exceeds 22, there is the tendency toward crystallization as time passes.

The emulsifier which is component (a) should preferably have relative ratios of (1), (2) and (3) within a polygon (including each side thereof) bounded by the 45 points A(25:73:2), B(25:55:20), C(70:10:20) and D(88:10:2) of the ternary composition diagram of the sole FIGURE. The emulsifier is used in an amount of 0.2 to 5 wt% (hereinafter referred to simply as %) of the cosmetic emulsion and preferably in an amount of 0.4 to 50 3%. Outside the range, the emulsification does not proceeds satisfactorily.

The oil component (b) should contain 90% or more of at least one silicone oil represented by the general formula (II), (III) or (IV). Aside from these silicone oils, 55 there may be used hydrocarbons such as liquid paraffin, paraffin wax, ceresin, squalane and the like, natural animal and plant oils such as olive oil, jojoba oil, mink oil and the like, and synthetic ester oils such as octyldodecyl myristate. These oils are used in an amount of 60.5 to 10%, preferably 1 to 5%, of the cosmetic emulsion. Over 10%, oiliness is unfavorably strengthened.

The aqueous phase (c) should have a ratio by weight of ethyl alcohol and water in the range of 50/50 to 2/98. The ratio is preferably in the range of 30/70 to 5/95. 65 When the ethyl alcohol/water ratio exceeds 50/50, the emulsification becomes unstable and the resulting emulsion is more stimulative against the skin. If, on the con-

trary, the ratio of ethyl alcohol/water is smaller than 2/98, refreshness does not appear. The aqueous phase may further comprise humectants such as glycerin, sorbitol, maltitol propylene glycol, dipropylene glycol, 5 1, 3-butylene glycol, sodium pyrrolidonecarboxylate, polyoxyethylene methylglucoside, polyoxypropylene methylglucoside, glucose and the like amino acids such as glycine, serine, proline and the like, and medical agents such as antiinflammatory agents, bactericides, vitamins and the like. The aqueous phase is used in an amount of 85 to 99.3%, preferably 92 to 98.6%, of the cosmetic emulsion.

The cosmetic emulsion of the invention is preferred to be low in viscosity and has generally a viscosity of below 100° centipoises at 25° C., preferably below 20 centipoises. Over 100 centipoises, the resulting emulsion is felt sticky upon application thereof.

The cosmetic emulsion of the invention is prepared by the procedure which comprises adding an ethyl alcohol aqueous solution, which has an ethyl alcohol/water ratio by weight of 20/80 to 70/30, to a mixture of (a) 0.2 to 5 parts by weight of an emulsifier and (b) 0.5 to 10 parts by weight of an oil under agitation at a temperature of from 20° to 45° C. to produce an o/w emulsion, and, if necessary, further adding water or an ethyl alcohol aqueous solution to give an intended composition.

In the practice of the invention, when the ethyl alcohol aqueous solution having an ethyl alcohol/water ratio by weight of 20/80 to 70/30, preferably 30/70 to 60/40, is added in an amount of 0.5 to 5 times by weight the mixture of (a) and (b), a gellike or highly viscous liquid one phase product is produced. Further addition of ethyl alcohol results in an o/w emulsion. If necessary, water or an ethyl alcohol aqueous solution is so added that the ratio by weight of ethyl alcohol/water is in the range of 50/50 to 2/98. The finally added water or ethyl alcohol aqueous solution should preferably be controlled at a temperature of from 5 to 45° C.

One of features of the invention resides in that one phase region is formed by emulsification using a high content of ethanol, through which region uniform and fine emulsified particles are obtained. The fact that fine emulsified particles are obtained through the one phase region during the course of the emulsification is known per se with regard to ordinary oil/emulsifier/water systems and oil/emulsifier/water/dihydric alcohol systems ("Journal of Chemical Society of Japan, 10, 1399 (1983), by Sagitani). The method for making cosmetic emulsions according to the invention is different from these known methods in that one phase region is formed under conditions of a high ethyl alcohol concentration in the system comprising an oil chiefly made of silicone oil, thereby obtaining a stable, low viscosity emulsion composition comprising fine, uniform emulsified particles. Moreover, it is also known from, for example, Japanese Laid-open Patent Application No. 58-131910 that polydiorganosiloxane-polyoxyalkylene copolymers are effective for emulsifying silicone oils. The present invention is also different from this prior art in that the emulsifier useful in the present invention consists of dimethylpolysiloxane-polyoxyalkylene copolymer, a surface active agent having an HLB value not smaller than 10, and a linear, saturated higher alcohol having from 12 to 22 carbon atoms in specific ratios, that an aqueous solution of the dimethylpolysiloxane-polyoxyalkylene copolymer has such a low clouding point of from 20° to 45° C. that it does not serve as an emulsifier under ordinary emulsifying conditions, but shows good performance as the emulsifier by defining the content of ethyl alcohol and the emulsification temperature in certain ranges, respectively.

The present invention is described by way of exam- 5 ples.

EXAMPLE 1

Cosmetic emulsions having the formulations indicated in Table 1 were prepared to check the emulsion 10 stability. The results are shown in Table 2.

TABLE 2-continued

•	State of Emulsion	(prese	Stability rved for 1 month)
В	good	40° C.	no change
		5° C.	no change
С	good	40° C.	no change
		5° C.	no change
Compara	ative Products:		-
D	rather poor	40° C.	separation
		5° C.	separation as crystals
E	separated immediately		

TABLE !

	Composition (%)								
	Proc	lucts of Inv	ention		Comparative Products				
Ingredients	A	В	С	D	E	F	G		
Oil Phase:	:	*	٠,						
Dimethylpolysiloxane- polyoxyalkylene copolymer (clouding point: 30° C.)	0.5	0.5	0.93	0.5	. –	0.5	0.81		
Dimethylpolysiloxane- polyoxyalkylene copolymer (clouding point: 80° C.)	· –	· .=	-	- .	0.5	_	. -		
POE(20) Sorbitan monooleate (HLB: 15)	0.5	0.4	0.12	0.2	0.5	_	0.28		
Sorbitan monooleate (HLB: 4.3)	· — .	-			_	0.5	– .		
Cetanol	0.1	0.08	0.05	0.4	0.1	0.1	0.01		
Dimethylpolysiloxane	1.5	2 .	1.5	1.5	1.5	1.5	1.5		
Methylphenylpolysiloxane Perfume	1.5 small amount	small amount	1.5 small amount	1.5 small : amount	1.5 small amount	1.5 small amount	1.5 small amount		
Alcohol Phase:									
Ethyl alcohol	10	20	10	. 10	10	10	10		
Water	10	25	10	10	10	10	10		
Glycerine Aqueous Phase:	3	2	3	. 3	3	3	3		
Water	balance	balance	balance	balance	balance	balance	balance		

Preparation: The oil phase was heated to 60° C. and all the ingredients were uniformly dissolved, followed by cooling down to 36° C. and keeping at a constant level. To the oil phase was added under agitation the alcohol phase which was controlled at 36° C. Thereafter, the aqueous phase of 25° C. was added to the mixture to give an intended composition.

TABLE 2

State of Emulsion	Stability (preserved for 1 month)
Products of Invention:	
A good	40° C. no change 5° C. no change

after preparation

F separated immediately
after preparation

G rather poor

40° C. separation 5° C. separation

As will be clear from Table 2, the cosmetic emulsions of the invention showed good stability.

EXAMPLE 2

Emulsion compositions of the formulations indicated in Table 3 were prepared to check the emulsion stability and viscosity. The results are shown in Table 4. The preparation of each emulsion was made in the same manner as in Example 1.

TABLE 3

	Composition (%)							
	Products of Invention		Comparative Products				s	
Ingredients	H	1	J	K	L	M	N	0
Oil Phase:	:	٠.						
Dimethylpolysiloxane- polyoxyalkylene copolymer (clouding point: 30° C.)	0.5	0.3	0.3	0.3	0.3	0.3	0.3	0.3
POE(20) hexadecyl ether (HLB: 16)	0.4	0.5	0.5	0.5	0.5	0.5	0.5	0.5
Cetostearyl alcohol	0.1	0.2	0.2	0.2	0.2	0.2	0.2	0.2
Methylphenylpolysiloxane	1.0	1.0	1.0	1 0. 0	1.0	1.0	1.0	1.0
Cyclic dimethylpolysiloxane	. 2.0	1.0	_ ` ` ` `	6.0	1.0	1.0	1.0	1.0
Octyldodecyl myristate	0.2	 .	1.0	_	_	_	:	
Perfume	small amount	small amount	small amount	small amount	small amount	small. amount	small amount	small amount
Alcohol Phase:							,	,
Ethyl alcohol	8	15	· 15	15	50	15		1.5
Water	12	15	15	15	30	3	15	5

TABLE 3-continued

				Compos	sition (%)				
	Products of Invention		Comparative Products						
Ingredients	Н	I	J	K	L	М	N	0	
Dipropylene glycol Aqueous Phase:	4	3	3	3	3	3	3	3	
Water	balance	balance	balance	balance	balance	balance	balance	balance	

TABLE 4

	Emulsion State (immediately after prepara- tion)	(pr	tability eserved 1 month)	Viscosity (25° C.) (cps.)	15
Prod	ucts of Invention:				_
H.	good	40° C.	no change	6	
	_	5° C.	no change		
1	good	40° C.	no change	4 .	
	_	5° C.	no change		
Com	parative Products:				20
J	good	40° C.	separation	4	
-	9	5° C.			
K	rather poor	40° C.		600	
		5° C.			
L	good	40° C.		3	
	·	5° C.	separation		25
M	separated immediately			_	
	after preparation				
N	separated immediately		<u> </u>	_	
	after preparation				
0	rather poor	40° C.	. separation	6	
	•	5° C.	separation		30

As will be clear from Table 4, the cosmetic emulsions of the invention showed good stability though very low in viscosity.

EXAMPLE 3

Five samples, in total, including the cosmetic emulsion A obtained in Example 1 and the cosmetic emulsion I of Example 2, both of which were the products of the invention, and the comparative emulsions J, K and O of Example 2 were organoleptically evaluated by 10 expert panelers with regard to the feelings to the touch. The results are shown in Table 5 below.

тм	RI	С	•
11	.DL	æ	J

	Refresh- ness	Sticki- ness	Oili- ness	Affinity	Appear- ance
Products	of Invention				
Α	1.6	-0.8	-1.2	+1.5	+1.8
1	2.0	-1.6	-1.9	+1.4	+1.6
Compara	ive Products:				
3	2.0	+1.1	-0.5	-0.2	+1.6
K	1.8	+0.8	+1.7	-1.4	-1.0
0	-0.8	-1.4	-0.8	+0.4	- 1.6

Evaluation Standard:								
	Refresh- ness	Sticki- ness	Oiliness	Affinity for Skin	Appear- ance			
+2	Very good	Very great	Very great	Very good	Very good			
+1	Rather good	Rather great	Rather great	Rather good	Rather good			
0	Moderate	Moderate	Moderate	Moderate	Moderate			
-1	Bad	Small	Small	Rather poor	Rather			
-2	Very bad	Very small	Very small	Very poor	Very poor			

(The values are average values of ten panelers.)

As will be seen from Table 5, the products of the invention are better in refreshness, less in stickiness and

oiliness, and better in affinity for skin and appearance than the comparative products.

EXAMPLE 4

Emulsion compositions P, Q of the invention having formulations indicated in Table 6, a comparative product R free of any oils, and commercially sold lotion S were each applied onto the flexor side of the forearm of men, followed by measurement of a skin conductance. The skin conductance corresponds to a constant of water in the epidermic corneum, meaning a greater amount of water at a higher conductance value. The measurement was effected using a high frequency impedance meter (Capacitance-conductance Meter Model 354, by IBS Co., Ltd.). The results are shown in Table 7 below.

TABLE 6

30		c	%)	
		Produ Inver	Compara- tive Product	
	Ingredients	P	Q	R
35	Oil Phase: Dimethylpolysiloxane- polyoxyalkylene copolymer	0.4	0.4	_
40	(clouding point: 28° C.) POE(60) hardened caster oil (HLB: 14.5)	0.9	0.9	0.9
40	Stearyl alcohol	0.1	0.1	_
	Dimethylpolysiloxane	2.5	2.5	
	Methylphenylpolysiloxane	1	1	_
	Perfume	small	small	small
		amount	amount	amount
	Alcohol phase:			
45	Ethyl alcohol	12	12	12
	Water	16	16	16
	Glycerine	4	_	4
	Aqueous phase:			
	Water	balance	balance	balance

(The comparative product R was a transparent liquid, not an emulsion.)

Preparation: Prepared in the same manner as in Example 1.

TABLE 7

	Time after Application of Tested Sample (Minutes)						
Tested Sample	10	30	60	90	120		
Products of Invention:							
P	360	320	298	316	300		
Q	336	293	288	280	288		
Comparative Products:	_						
R	170	100	98	95	92		
S	133	91	88	87	83		
(Commercial available lotion)							
Control (nothing	82	82	80	83	80		

50

55